Semester	VI
Course	Major
Paper Code	
Paper Title	Atomic and Molecular Physics & Electronics III
No. of Credits	4
Theory / Practical / Composite	Theory
Minimum No. of preparatory hours per week a student has to devote	4
Number of Modules	2
Syllabus	Module A - Atomic and Molecular Physics [24 L]  Brief review of the hydrogen atom: energy levels and quantum numbers  Atoms in Electric & Magnetic Fields: Electron angular
	momentum. Space quantization. Electron Spin and Spin Angular Momentum. Larmor's Theorem. Spin Magnetic Moment. Stern-Gerlach Experiment. Basic Stark effect; selection rules (statement & simple illustrative examples).
	[10 Lectures]
	Multi-Electron Atoms and Angular Momentum Coupling: Spectrum of alkali atoms; Pauli exclusion principle, Symmetric and Anti-symmetric wavefunctions, Singlet and Triplet structure; Hund's rule, L-S (Russell-Saunders) coupling scheme, brief note on J-J coupling; Spectroscopic symbols for simple configurations: transition rules; qualitative introduction to electron-electron interactions
	[5 Lectures]
	Fine Structure and Zeeman Patterns: Spin-orbit coupling: interaction between electron spin and orbital angular momentum, spin-orbit splitting patterns; fine structure: Landé g-factor, normal and anomalous Zeeman splitting patterns; qualitative overview of Paschen-Back (strong-field) behavior; electric dipole (E1) selection rules in term notation; brief remarks on line strengths and intensity patterns
	[4 Lectures]
	Molecular Spectra of Diatomic Molecules: Franck-Condon principle; Electronic, Vibrational and Rotational spectra of diatomic molecules; Selection rules for di-atomic transitions; Qualitative comparison of atomic vs. molecular energy scales

(binding, vibrational, and rotational energies); Introduction to diatomic molecular structure and energy level spacing

[5 Lectures]

## **Module B - Electronics III**

[24 L]

**Junctions and devices:** Charge, Electric field, potential variation across the transition region of a P N junction. Metalsemiconductor diode: Device structure and energy band diagram, Schottky effect, barrier height, ohmic contact.

Field-Effect Transistors: JFET: Device structure and operation, pinch-off and saturation, I-V characteristics. MOS capacitors: Energy band diagram, accumulation, depletion and inversion mode of operation, threshold voltage, flat band voltage, capacitance-voltage characteristics, MOSFET: Device structure and operation, band diagram, I-V characteristics, Short channel effects.

## [10 Lectures]

Analog communication: Principles and Applications: Introduction to analog communication, Signals and Spectral Analysis, Amplitude Modulation (DSB, DSB-SC, SSB), Frequency and Phase Modulation, Noise in Analog Communication, Analog Communication system blocks – basic idea of receivers, transmitters, RF amplifiers, Mixers, filters. Application in Physics and Instrumentation.

## [9 Lectures]

## **Digital Communication:**

Introduction to digital communication, Sampling and Pulse Code Modulation (PCM), Line Coding (NRZ, RZ, Manchester) and digital communication techniques (ASK, FSK, PSK), Noise, SNR, Error Detection, Application in Physics.

[5 Lectures]

Learning Outcomes	Module A
	CO1: In this course, first some applications of quantum mechanics for simple systems will be dealt with. This will initiate the idea of how a theoretical framework is applicable to real life problems.
	CO2: At the last section of this course, more advanced applications of quantum mechanics will be discussed which arise in atomic physics, nuclear physics and in other areas of physics
	CO 3: Students will be able to explain spectra of di-atomic molecules and will be able to extract information regarding bond-length, anharmonicity constant, dissociation energy from the spectra.
	Module B
	<ol> <li>Understanding the basics in junctions formed by metal, semiconductor and insulators.</li> <li>Understanding how voltage dependent transistors work.</li> <li>getting an overview of the pros and cons of MOSFET</li> <li>Basic concepts of modulation and demodulation</li> <li>Understanding the requirements and limitations of analog and digital modulation</li> </ol>
Reading/Reference Lists	Module A References
	<ol> <li>D.J. Griffiths &amp; D.F. Schroeter, Introduction to Quantum Mechanics, CUP</li> <li>B.H. Bransden &amp; C.J. Joachain, Physics of Atoms and Molecules, Longman</li> <li>R. Eisberg &amp; R. Resnick, Quantum Physics of Atoms, Molecules, Solids, Nuclei and Particles, Wiley</li> <li>C.J. Foot, Atomic Physics, OUP Oxford.</li> <li>P. Atkins &amp; R. Friedman, Molecular Quantum Mechanics, OUP Oxford.</li> <li>G. Herzberg, Molecular Spectra and Molecular Structure: I. Spectra of Diatomic Molecules, Springer US</li> <li>P.M. Mathews &amp; K. Venkatesan, A Textbook of Quantum Mechanics, Tata McGraw-Hill Publishing</li> <li>H.E. White, Introduction to Atomic Spectra, McGraw-Hill book Company, Incorporated.</li> <li>Arthur Beiser, Concepts of Modern Physics, McGraw-Hill Education (India) Pvt Limited.</li> <li>S. Gasiorowicz, Quantum Physics, Wiley India Pvt. Limited.</li> </ol>
	Module B References:
	1. Streetman and Banerjee: Solid State Electronic Devices – Prentice Hall India
	2. S M Sze and Kwok K Ng: Physics of semiconductor devices – Wiley publishers

	<ol> <li>Roddy and Coolen: Electronic Communications, Pearson</li> <li>Taub and Schilling: Principles of Communication Systems – McGraw Hill</li> <li>Kennedy and Davis: Electronic Communication Systems – Tata McGraw Hill</li> <li>B P Lathi: Modern Digital and Analog communication Systems – Oxford</li> <li>Millman and Halkias: Integrated Electronics - McGraw-Hill</li> </ol>
Evaluation	CIA 30 (2 x 10 + 5 Assignment +5 Attendance) SEM Exam 70
Paper Structure for Theory Semester Exam	For each module of 35 Marks: 15 marks from 3 mark questions (5 / 7) 20 marks from 10 mark questions (2 / 3)